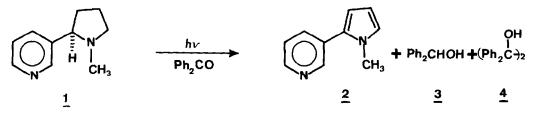
Photodehydrogenation of Substituted Pyrrolidines Hydrogen Transfer Induced by Benzophenone

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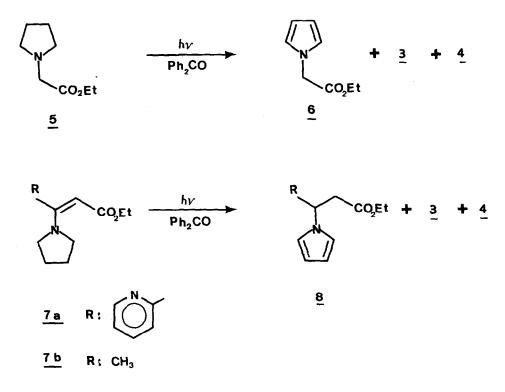
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When excited in their ${}^{3}(n, \pi^{*})$ excited states, ketones are able to abstract hydrogen from carbon atoms bearing heteroatom substituents. Several groups have studied the photochemical oxidation of amines (1) and amides (2) in the presence of benzophenone and found that aromatization of 5-membered ring amines occurs under these conditions. For example, when nicotine is irradiated in the presence of dyes, the pyrrolidine moiety is oxidized to a pyrrole (3a). Similarly, photolysis of imidazolines in acetone produces dehydrogenation and formation of the corresponding imidazoles (3b).

Consistent with these observations, we have found that irradiation at 254 nm of a benzene solution of nicotine $(10^{-2}M)$ containing benzophenone $10^{-1}M$ provides nicotyrine 2 (40%), benzhydrol 3, benzopinacol 4, and some uncharacterized polymeric material. Similarly, irradiation of 5 under identical conditions gives 6 (4). In neither of these cases were any products detected which would have resulted from coupling of benzhydryl and α -pyrrolidino radicals.

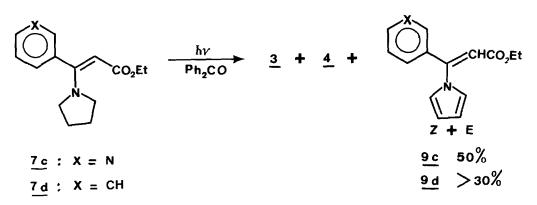


In striking contrast to these results, irradiation at 254 nm (or 366 nm) of a solution of $\underline{7a}$ in benzene (or acetonitrile) containing benzophenone leads to $\underline{3}$, $\underline{4}$ and $\underline{8a}$ (50%) (5). Likewise, $\underline{8b}$ (30%) is the major product obtained from $\underline{7b}$ in similar conditions. Thus, in these cases oxidation of the pyrrolidino substituent is accompanied by reduction of the conjugated carbon-carbon double bond (6).

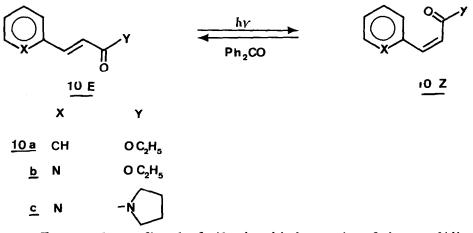


It is well known that excited pyridines, quinolines (7) and carboxylic esters (8) are able to abstract hydrogen atoms. In order to determine if <u>8</u> is obtained from the triplet excited state of <u>7</u> or from a chemically induced sensitization, we have irradiated <u>7a</u> in the presence of <u>p</u>-methoxyacetophenone ($E_T \approx 74$ kcal) a photosensitizer whose ability to abstract hydrogen is very low (9). Since <u>7a</u> is recovered almost quantitatively from this experiment, we conclude that <u>8a</u> results from chemical sensitization by benzophenone.

This hydrogen transfer process seems to be very sensitive to molecular structure. For example when $\underline{7c}$ or $\underline{7d}$ are irradiated in the presence of benzophenone the conjugated double bond in $\underline{7}$ is not reduced and $\underline{3}$, $\underline{4}$ and $\underline{9}$ are the main products of the reaction. However, in order to effect both oxidation of the pyrrolidine moiety and reduction of the acrylic group in this reaction, it does seem necessary for the pyrrolidine group to be β to the ester substituent. Thus, irradiation of $\underline{10}$ in the presence of benzophenone only



leads to $Z \neq E$ isomerization following energy transfer from the sensitizer. Furthermore, no reduction is observed when <u>10a</u> is irradiated in N-methylpyrrolidine in the presence of an aromatic ketone.



These results confirm the facile photodehydrogenation of the pyrrolidino substituent and show that hydrogen transfer can be induced by benzhydryl radicals. The mechanistic details of these reactions are currently under investigation.

Acknowledgements

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